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elements. (Section 2.5) Predict whether a bond between two atoms of different elements would ... To indicate that a chemical reaction requires the continuous addition of heat in order to proceed, we place an upper case Greek delta,  $\Delta$ , above the arrow in the equation.

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3) Single-Replacement Reactions  $\hat{\neq}$  A chemical change where one element replaces a second element in a compound.  $\hat{\neq}$  It is identifiable by its symmetry. On the reactant side there is a single element and a compound, as there is on the product side.  $\hat{\neq}$  No reaction will occur if the most reactive element is already in the compound

### The 5 Types of Chemical Reactions (Chapter 11)

Chemical Reactions-Multiple Choice Review ... In a word equation representing a chemical reaction, the reactants are written on the left and the products on the right. 14) Chemical equations \_\_\_\_\_. ... constituent elements. A)  $\text{PbO}_2 \text{Pb} + 2\text{O}$  B)  $\text{PbO}_2 \text{Pb} + \text{O}_2$  C)  $\text{Pb}_2\text{O} 2\text{Pb} + \text{O}$

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